

## VIBRATIONAL SPECTRA OF 2,6-DICHLOROANISOLE

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The infrared absorption spectra of 2,6-dichloroanisole has been recorded by a Perkin-Elmer-521 spectrophotometer in the region  $250\text{--}4000\text{ cm}^{-1}$  on a thin film. The spectra have been analysed assuming  $C_{2v}$  point group for the molecule. A tentative assignment of the observed bands to different fundamental modes has been made. The shift of the C-Cl and Caryl-O frequencies are discussed.

*1. Introduction*

Vibrational spectra of anisole and its monosubstituted derivatives have been studied extensively by many workers [1-7], but very little work appears on di-substituted anisoles [8-10]. In order to extend this study to di-substituted anisoles and to study the effect of substitution, the molecule 2,6-dichloroanisole has been undertaken at present.

*2. Experimental*

The spec-pure chemical, 2,6-dichloroanisole, obtained from M/s FLUKA AG, Switzerland, was used as supplied without further purification. The infrared absorption spectra have been recorded by a Perkin-Elmer-521 spectrophotometer on a thin film in the region  $250\text{--}4000\text{ cm}^{-1}$ . The spectrometer was calibrated by running the spectra of a thin sheet of polystyrene.

*3. Results and discussion*

The infrared absorption spectra of 2,6-dichloroanisole is shown in Fig. 1 and the observed frequencies along with their probable assignments are presented in Table I. Wilson's [11] notations have been followed in Table I, where *a* and *b* denotes the compo-

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TABLE I

Assignment of vibrational frequencies of 2,6-dichloroanisole

Symmetry species	Vibration number	Position of the bands along with their visual intensities* [cm <sup>-1</sup> ]	Description of mode
$A_1$	2	3065 (mb)	C—H stretch
	20a	3045 (vw)	C—H stretch
	8b	1590 (ms)	C—C stretch
	19b	1475 (vs)	C—C stretch
	13	1260 (vvs)	C <sub>aryl</sub> — O stretch
	9a	1180 (s)	C—H i.p.b.
	7a	1155 (s)	C—Cl stretch
	12	1002 (vvsb)	ring planar defor., trigonal bending
	1	780 } 792 } (vvs)	C—C stretching, breathing type
	6a	612 (vvs)	C—C i.p.b.
	18b	355 (s)	C—Cl i.p.b.
$B_2$	20b	3005 (mw)	C—H stretch
	8a	1570 (vvs)	C—C stretch
	19a	1422 (vs)	C—C stretch
	14	1332 (s)	C—C stretch, Kékulé vib.
	7b	1100 (vs)	C—Cl stretch
	3	1290 (m)	C—H i.p.b.
	9b	1208 (vvs)	C—H i.p.b.
	6b	560 (s)	C—C i.p.b.
	15	410 (ms)	C—OCH <sub>3</sub> i.p.b.
	18a	325 (vw)	C—Cl i.p.b.
$A_2$	17a	820 (mb)	C—H o.p.b.
	16a	550 (ms)	C—C o.p.b.
	10a	—	—
$B_1$	4	712 (s)	C—C o.p.b.
	17b	910 (s)	C—H o.p.b.
	11	750 (vvs)	C—H o.p.b.
	16b	402 (ms)	C—C o.p.b.
	10b	—	—
	5	272 (m)	C—OCH <sub>3</sub> o.p.b.
OCH <sub>3</sub> group vibrations		2945 (vs) 2860 (vs) 2825 (s) 1447 (s) 1390 (sh)	C—H asym. stretch C—H asym. stretch C—H sym. stretch CH <sub>3</sub> asym. defor. CH <sub>3</sub> sym. defor.

TABLE I (continued)

	1045 (ms)	Calkyl—O stretch
	1072 (vvs)	CH <sub>3</sub> rocking
	1035 (ms)	CH <sub>3</sub> rocking
	260 (w)	C—O—C i.p.b.

Other vibrations

590 (mw) = 272 + 325, 1515 (mw) = 792 + 712, 1595 (mw) = 2 × 792, 1617 (vw) = 792 + 820, 1655, 1662 (m) = 1045 + 612 or 1100 + 560, 1675 (m) = 1260 + 410, 1692 (mw) = 910 + 780, 1790 (m) = 1035 + 750, 1802 (m) = 1045 + 750, 1860 (ms) = 1045 + 820, 1930 (ms) = 1155 + 780, 2060 (ms) = 2 × 1035, 2225 (mw) = 1155 + 1072, 2270 (m) = 1260 + 1002, 2510 (mb) = 2 × 1260, 2930 (mw) = 1475 + 1447, 3535 (m) = 2825 + 712.

\* Intensities are shown in parentheses against each wave number. i.p.b. → in-plane bending and o.p.b. → out-of-plane bending:

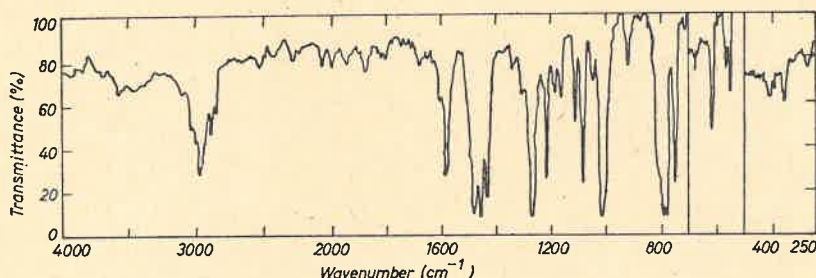


Fig. 1. Infrared absorption spectra of 2,6-dichloroanisole

nents of doubly degenerated modes which splits into their components when symmetry reduces by substitution in the benzene molecule.

Assuming the OCH<sub>3</sub> group to be a single mass point, the molecule 2,6-dichloroanisole would belong to the C<sub>2v</sub> point group symmetry under which 30 normal modes will be distributed as 11a<sub>1</sub> + 3a<sub>2</sub> + 6b<sub>1</sub> + 10b<sub>2</sub>. In addition to these, there will also appear 12-vibrations due to the OCH<sub>3</sub> group. In the absence of Raman lines with depolarisation data and the vapour phase infrared spectra, the present assignments are made on the basis of visual intensities and data available for similar molecules. For the present case some important assignments are discussed below:

### 3.1. A<sub>1</sub> species

The present molecule, 2,6-dichloroanisole, is a tri-substituted benzene, thus, three C—H stretching vibrations,  $\nu_2$  and  $\nu_{20a}$  belonging to the A<sub>1</sub> species and  $\nu_{20b}$  belonging to B<sub>2</sub> species are expected in the region 3000–3100 [12]. The vibrations 2 and 20a have been identified at 3065 and 3045 cm<sup>-1</sup> respectively in the present molecule. In tri-substituted

benzenes [13] the frequency of one of the two ring modes  $\nu_1$  and  $\nu_{12}$  decreases to  $820\text{ cm}^{-1}$  while the other remains around  $1000\text{ cm}^{-1}$ . Thus the bands observed at  $780$  and  $1002\text{ cm}^{-1}$  in the present study have been assigned to the  $\nu_1$  and  $\nu_{12}$  vibrations respectively. These are also in agreement with the assignments made by Goel et al. [8, 14], Srivastava [15] and Pandey and Singh [16] for chloro-substituted benzenes.

In the present case, the  $\nu_{8b}$  and  $\nu_{19b}$  vibrations belonging to the  $A_1$  species and which represents the C—C stretching vibrations, have been identified at  $1590$  and  $1475\text{ cm}^{-1}$  respectively. These assignments are in agreement with literature values [14, 17].

### 3.2. $B_2$ species

C—H stretching vibration  $\nu_{20b}$  belonging to the  $B_2$  species has been identified at  $3005\text{ cm}^{-1}$  in the present molecule. The vibrations  $\nu_{8a}$  and  $\nu_{19a}$  corresponding to ring stretching and belonging to the  $B_2$  species, have been identified at  $1570$  and  $1422\text{ cm}^{-1}$  respectively.

As a result of substitution at positions 1, 2 and 6 of the ring, the stretching vibrations  $\nu_{13}$ ,  $\nu_{7a}$  and  $\nu_{7b}$  become X-sensitive. Bishui [18] assigned the C—Cl stretching mode in 2,6-dichloroaniline at  $1050$  and  $1070\text{ cm}^{-1}$  while Scherer and Evans [19] assigned this mode at  $1070$  and  $1152\text{ cm}^{-1}$  in m-dichlorobenzene. Thus the bands at  $1155$  and  $1100\text{ cm}^{-1}$  have been assigned to the C—Cl stretching mode ( $\nu_{7a}$  and  $\nu_{7b}$  respectively) in the title compound. This also finds support from the literature values [20–22].

In the infrared spectra of aryl and alkyl ethers, several workers [1, 4, 8] have reported two stretching vibrations involving the oxygen atom, viz.,  $C_{\text{aryl}}\text{—O}$  and  $C_{\text{alkyl}}\text{—O}$  stretching vibrations. The  $C_{\text{aryl}}\text{—O}$  stretching mode was observed at  $1266 \pm 6\text{ cm}^{-1}$ , but in anisole due to the hyper-conjugation effect of the methyl group, it is observed at a higher frequency. The  $C_{\text{alkyl}}\text{—O}$  stretching vibration occurs at  $1035 \pm 10\text{ cm}^{-1}$  in anisoles. In the present molecule the band observed at  $1045\text{ cm}^{-1}$  has been assigned to the  $C_{\text{alkyl}}\text{—O}$  stretching while the  $\nu_{13}$  vibration corresponding to  $C_{\text{aryl}}\text{—O}$  stretching has been identified at  $1260\text{ cm}^{-1}$ . These also find support from the literature values [5, 6, 10].

$\nu_3$  and  $\nu_{9b}$  vibrations which correspond to the C—H in-plane bending modes have been identified at  $1290$  and  $1208\text{ cm}^{-1}$  in the title compound, while C—C in-plane bending mode  $\nu_{6b}$  (corresponding to  $e_{2g}608$  mode of benzene) have been identified at  $560\text{ cm}^{-1}$ .

### 3.3. $A_2$ species

The out-of-plane C—Cl stretching mode  $\nu_{10a}$  of the  $A_2$  species and  $\nu_{10b}$  of the  $B_1$  species lie below the region under study, thus they could not be observed. Since the vibrations belonging to the  $A_2$  species are infrared inactive, they will either be absent or appear weakly. Thus a band with medium intensity at  $820\text{ cm}^{-1}$  has been assigned to the  $\nu_{17a}$  mode and that at  $550\text{ cm}^{-1}$  to the  $\nu_{16a}$  mode belonging to the  $A_2$  species.

### 3.4. $B_1$ species

The band corresponding to the  $b_{2g}$  (703) mode of benzene ( $\nu_4$  vibration) has been identified at  $712\text{ cm}^{-1}$ . The  $\nu_{17b}$  and  $\nu_{11}$  vibrations corresponding to C—H out-of-plane bending modes have been assigned at  $910$  and  $750\text{ cm}^{-1}$  in the present case.



### 3.5. $\text{OCH}_3$ group vibrations

There are three C—H stretching vibrations in the methyl group, one of these is symmetric while the other two are asymmetric. The C—H asymmetric stretching vibrations have very nearly the same magnitudes and usually appear with varying intensity in the region  $3000\text{--}2900\text{ cm}^{-1}$ . In view of this, the bands observed at  $2945$  and  $2860\text{ cm}^{-1}$  have been assigned to the C—H asymmetric stretching and that at  $2825\text{ cm}^{-1}$  to C—H symmetric stretching vibrations. This finds support from the work of Mooney [21] on halogenated toluenes and the work of Goel et al. [8, 10] and Dwivedi and Sharma [5] on substituted anisoles. Further, for one methyl group attached to the benzene ring, two rocking modes are expected. Wilmshurst and Bernstein [23] have assigned these modes in toluene at  $1040$  and  $1080\text{ cm}^{-1}$ . In view of this, the  $\text{CH}_3$  rocking modes in the present case have been assigned at  $1035$  and  $1072\text{ cm}^{-1}$ . This is also in agreement with the values suggested in the literature [5, 10, 24].

### 3.6. Shift of C—Cl and C— $\text{OCH}_3$ frequencies

It is interesting to note the shift of C—Cl and  $\text{C}_{\text{aryl}}\text{—O}$  stretching frequencies in the present case compared to that in 2,3-dichloroanisole [10]. The two C—Cl stretching frequencies and  $\text{C}_{\text{aryl}}\text{—O}$  stretching mode were observed at  $1171$ ,  $1119$  and  $1273\text{ cm}^{-1}$  in 2,3-dichloroanisole while in the present case at  $1155$ ,  $1100$  and  $1260\text{ cm}^{-1}$  respectively. In 2,6-dichloroanisole both chlorine atoms are nearer to the  $\text{OCH}_3$  group than in 2,3-dichloroanisole. Since chlorine is electronegative, both Cl atoms, which are nearer to the  $\text{OCH}_3$  group in 2,6-dichloroanisole, may perturb the  $\text{OCH}_3$  group and C— $\text{OCH}_3$  frequencies more than in 2,3-dichloroanisole. According to Nakanishi and Solomon [25] such a perturbation may arise from mesomeric and inductive effects. In turn, this also explains the shift of C—Cl stretching frequencies. This finds support from the similar shift observed in 2,4- and 2,6-dibromophenols [26].

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